Tetrahedron Letters No.42, pp. 3715-3718, 1970. Pergamon Press. Printed in Great Britain.

SYNTHESES <u>VIA</u> DIHYDRO-1, 3-OXAZINES. XI. A SYNTHESIS OF FUNCTIONALLY SUBSTITUTED ALDEHYDES A.I. Meyers, G.Ray Malone, and H. Wayne Adickes Department of Chemistry Louisiana State University in New Orleans Lake Front New Orleans. Louisiana 70122

(Received in USA 15 May 1970; received in UK for publication 10 August 1970)

The use of dihydro-1,3-oxazines in synthesis has been demonstrated with respect to obtaining aldehydes,¹ ketones,² and acids³ of varied structure. We now report our preliminary results regarding the formation of halo, cyano, hydroxy, and keto-substituted aldehydes along with dialdehydes by simple manipulation of the dihydro-1,3-oxazine system and readily available materials.

The lithic salt ¹ of the 2-methyldihydro-1, 3-oxazine <u>1</u> (R=H) was converted to the chlorobutyl derivative <u>2</u> (R=H, n=3, X=Cl) by treatment with 1-chloro-3-bromopropane at -78°. The crude product (92%) was reduced ¹ without purification to the corresponding tetrahydro-1, 3-oxazine (THO) <u>3</u> using aqueous sodium borohydride, which was cleaved (oxalic acid) to 5-chloropentanal <u>4</u>a in 51% yield (Table I). In a similar fashion the 2benzyldihydro-1, 3-oxazine, through its lithic salt <u>1</u> (R=Ph) was monoalkylated with 1, 4dibromobutane in 94% crude yield to <u>2</u> (R=Ph, n=4, X=Br) which was reduced and cleaved to 6-bromo-2-phenylhexanal <u>4</u>b in 60% yield. This sequence adequately exemplifies the utility of this method in obtaining ω -haloaldehydes of various chain lengths.

The selectivity of the lithio salt <u>1</u> (R=Ph) toward substituted halides (5-bromovaleronitrile) was illustrated by a 92% conversion to <u>2</u> (R=Ph, n=4, X=CN) without appreciable attack at the other reactive centers. The usual reduction with sodium borohydride followed by acid cleavage of <u>3</u> afforded 6-cyano-2-phenylhexanal (<u>4</u>c). Attempts to duplicate this sequence employing <u>1</u> (R=H) resulted in proton abstraction, addition to the cyano group, along with bromide displacement. The poor selectivity of the anion of the 2-methyl-DHO was

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overcome by a slight modification in procedure discussed later $(7 \longrightarrow 8a \longrightarrow 8b)$.

The synthesis of dialdehydes followed from this scheme by utilizing 2 eq of the lithio salt <u>1</u> (R=Ph) and 1 eq of 1, 4-dibromobutane to form the $a, \omega - \underline{bis}$ (dihydro-1, 3-oxazinyl) alkane <u>5</u> (R=Ph, n=4). Reduction with the requisite amount of borohydride solution cleanly reduced the C=N link in both oxazine rings (<u>6</u>) which was then hydrolyzed in oxalic acid to 2, 7-diphenyloctanedial (<u>4</u>d).



To further demonstrate the wide scope of this method for obtaining substituted aldehydes, a scheme starting with 2, 4, 4, 6-tetramethyl-5, 6-dihydro-1, 3-oxazine (DHO-CH₃) $\underline{?}$ and 1, 5-dibromopentane was investigated. The 6-bromohexyl derivative, $\underline{8}a$ (90%, 1658 cm⁻¹) was transformed into the corresponding nitrile $\underline{8}b$ (95%, 2240, 1658 cm⁻¹) by treatment with 1.8 eq sodium cyanide in DMSO (55°, 2 hr). This procedure, therefore, overcomes the poor selectivity of $\underline{1}$ (R=H) with halonitriles. Reduction and cleavage of $\underline{8}b$ gave 7-cyano-heptanal $\underline{9}a$ (Table 2). The bromohexyl oxazine $\underline{8}a$ could also be converted to the Grignard reagent $\underline{8}c$

9a (Table 2). The bromohexyl oxazine $\underline{8}a$ could also be converted to the Grignard reagent $\underline{8}c$ (THF, 66° , 3 hr)³ treated with deuterium oxide, and ultimately converted to 7-deuterioheptanal 9b. The Grignard reagent $\underline{8}c$ was also treated with benzaldehyde (MgBr₂) producing the alcohol $\underline{8}e$ (55%, 3225, 1658 cm⁻¹) which subsequently gave the 8-hydroxy-8-phenylaldehyde 9c. Reaction of $\underline{8}c$ with ethyl chloroformate proceeded normally giving rise to the 6-carbethoxy oxazine $\underline{8}g$ (40%, 1660, 1735 cm⁻¹). The latter was formed in higher yield (83%) from the nitrile $\underline{8}b$ using ethanolic hydrogen chloride. The successful introduction of the carboethoxy group is significant due to the fact that the anion of 7 (1 R=H) is poorly selective in its reaction with ω -bromoesters. Finally, the addition of phenylmagnesium bromide to the 6-cyanohexyl oxazine, $\underline{8}h$, produced the 6-benzoylhexyl derivative, $\underline{8}f$ (90%, mp 28-30, 1675, 1658 cm⁻¹). The C=N link was reduced (-45°) without affecting the carbonyl group and the tetrahydro-1, 3-oxazine hydrolyzed to the ketoaldehyde 9d.

Although the transformations described are of the standard type, they are reported here to illustrate the unique stability of the dihydro-1, 3-oxazine toward a variety of conditions.

<u>ACKNOWLEDGMENTS:</u> Financial assistance from the National Science Foundation (GP-9592), Lithium Corporation of America, Hoffmann-La Roche, and the Petroleum Research Fund, administered by the American Chemical Society, is gratefully acknowledged.

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Compound	R	n	x	% yield ^a	ir (cm ⁻¹)	δppm(CHO)	2,4-DNP
<u>4</u> a	Н	3	C1	51	1722	9.76	106-108 ^b
<u>4</u> b	Ph	4	Br	60	1725	9.63	136-138
<u>4</u> c	$\mathbf{P}\mathbf{h}$	4	CN	71	1720, 2243	9.66	16-117
<u>4</u> d	\mathbf{Ph}	4	PhCHCHO I	63 ^C	1716	9.61	210-213

TABLE I. Substituted Aldehydes $(\underline{4})$ - OHC - CH(CH₂)_nX

a) All new compounds gave satisfactory elemental and/or mass analyses; b) Y. Ban and T. Orshi, Chem. Pharm. Bull., <u>11</u>, 446 (1963); c) Mp 37-39⁰

TABLE II.	Substituted	Heptaldehydes	(<u>9</u>)

Compound	X	% yield	ir (cm ⁻¹)	δppm (CHO)	2,4-DNP
<u>9</u> a	CN	47	1718, 2240	9.71	76-77 ^a
<u>9</u> ъ	D	55 ^b	1724, 2178	9. 73	108-109
<u>9</u> c	PhCHOH	44	1720, 3400	9.70	
<u>9</u> d	PhC=O	49	1720, 1680	9.69	84-85 ^c

a) M. Ohno, N. Naruse, S. Torimitsu, I. Teresawa, J. Am. Chem. Soc., <u>88</u>, 3168 (1966);

b) Mass spectral analysis indicated that deuteriation was quantitative; c) Mp of corresponding keto acid [T. Weil and D. Ginsburg, J. Chem. Soc., 1291 (1957)], which was obtained by oxidation with chromic acid.

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